

89-259866/36 KAO CORP 22.01.88-JP-012061 (27.07.89) C11d-01/28 D061-01/04 Detergent compsn. for dry cleaning - contg. mixt. of sulpho- tri:carb:allylic ester salts C89-115536	D25 E19 G04 KAOS 22.01.88 *JO 1188-595-A	D(11-A1A, 11-A1B2, 11-B16, 11-D1) E(10-A9B8) G(4-B8)
<p>A new detergent compsn. for dry cleaning contains one or a mixt. of sulphotrichallylic ester salts of formulae (I) and (II)</p> $\begin{array}{c} \text{CH}_2\text{COOR}_1 \\ \\ \text{XO}_3\text{S}-\text{C}-\text{COOR}_2 \quad (\text{I}) \\ \\ \text{CH}_2\text{COOR}_2 \end{array} \quad \begin{array}{c} \text{XO}_3\text{S}-\text{CHCOOR}_1 \\ \\ \text{CHCOOR}_2 \quad (\text{II}) \\ \\ \text{CH}_2\text{COOR}_3 \end{array}$ <p>R_1, R_2 and R_3 = 1-20C opt. branched alkyl or alkenyl, or one of H, alkali metals, alkaline earth metals and org. amines (at least two of them are the alkyl or alkenyl gps.); and X = alkali metal alkali earth metal, or org amine.</p>	<p><u>USE/ADVANTAGE</u> The compsn. has good phase stability and a good water-solubilising ability (a hydrated ability). It thus readily forms a uniform pretreating liq with any commercial petroleum solvent.</p> <p><u>EXAMPLE</u> (I) and (II) include the sodium salts of tri-2-ethyl butyl sulphotrichallylate, trihexyl sulphotrichallylate and tri-2-ethyl hexyl sulphotrichallyate, and the potassium salts of trioctyl sulphotrichallylate, trioctyl sulphotrichallylate, and tributyl sulphotrichallylate. (6ppW31MBDwgNc 0/0).</p>	J00188595-A